

B.Tech II Year I Semester (R13) Supplementary Examinations June 2017 THERMODYNAMICS

(Mechanical Engineering)

Time: 3 hours

1

5

Max. Marks: 70

PART – A

(Compulsory Question)

Answer the following: (10 X 02 = 20 Marks)

(a) Convert 35°C into K and 60°F into R.

- (b) What is meant by quasi static process?
- (c) What is meant by pmm1?
- (d) Mention some applications in which steady flow equations are used.
- (e) Define the term entropy.
- (f) State Clausius theorem.
- (g) What is the important vapour process?
- (h) What is meant by joule Thomson coefficient?
- (i) Define Avogadro's law.
- (j) State Dalton's law of mixtures.

PART – B

(Answer all five units, $5 \times 10 = 50$ Marks)

A gas undergoes a quasi static expansion which follows P = a + bV, where a and b are constants. The initial and final volumes are 0.2 m³ and 1.2 m³. The initial and final pressures are 1000 kPa and 200 kPa respectively. The specific internal energy of the gas is given by the equation U = 1.5PV - 85 kJ/kg, where P is in kPa and U is in m³/kg. Calculate the net heat transfer.

OR

3 The various process of a gas undergoing following thermodynamic cycle:

Process 1-2 constant pressure P = 2.5 bar, $v_1 = 0.4 \text{ m}^3$, $W_{12} = 13 \text{ kJ}$

Process 2-3 compression PV = Constant, $U_3 = U_2$

Process 3-1 constant volume $U_1 - U_3 = -30 \text{ kJ}$

Calculate the net work for the cycle in kJ, heat transfer for process 1-2.

UNIT – II

- 4 A nozzle is a device for increasing the velocity of a steadily flowing stream. At the inlet to a certain nozzle, then the enthalpy of fluid passing is 3000 kJ/kg and velocity is 60 m/s. at the discharge end, the enthalpy is 2762 kJ/kg. the nozzle is horizontal and there is negligible heat loss from it.
 - (i) Find the velocity at the exit from the nozzle.
 - (ii) If the inlet area is 0.1 m² and the specific volume at inlet is 0.187 m³/kg, find the mass flow rate.
 - (iii) If the specific volume at nozzle exit is 0.498 m³/kg, find the exit area of the nozzle.

OR

In a steady flow apparatus, 120 kJ of work is done by each kg of fluid. The specific volume of the fluid, pressure and velocity at the inlet are 0.2 m³/kg, 550 kPa and 13 m/s. The inlet is 28 m above the floor, and the discharge pipe is at the floor level. The total heat loss between the inlet and discharge is 5 kJ/kg of fluid. In flowing through this apparatus, how much does the internal energy increases or decreases. The discharge condition are 0.5 m³/kg, 90 kPa and 250 m/s.

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(UNIT – III)

- 6 (a) A food freezer maintains temperature of -13°C. If heat leaks into the freezer at the continuous rate of 1.65 kJ/s. What is the power necessary to pump out heat continuously?
 - (b) 3 kg of air is expanded at constant pressure to three times its initial volume. Calculate the change in entropy if the initial temperature of the air is 300°C.

OR

- 7 (a) 2 kg of air at pressure P₁ and temperature 978 K is mixed with 2 kg of air at the same pressure but at 486 K. Determine the loss in availability if the atmospheric temperature is 310 K.
 - (b) 2 kg of air is compressed polytropically from pressure 2 bar and temperature 350 K to a pressure of 8 bar and 400 K. Determine the irreversibility and reversibility measure assuming the sink temperature of 348 K.

UNIT – IV

- 8 (a) Find the condition of the steam leaving the turbine and the work done/kg of the steam passing through the turbine if the steam is at 12 bar and 300°C is throttled till its pressure becomes 7 bar and it expands isentropically passing through a turbine until pressure falls to 1 bar. Also find the work done/kg of stem if the steam is directly passed through the turbine without throttling.
 - (b) Using Clausius clapeyron equation, estimate the enthalpy of vaporization at 220 °C saturation temperature. Take the following data $T_s = 220$ °c, $v_g = 0.086$ m³/kg, $v_f = 0.001109$ m³/kg, (dP/dT) = 52 kPa/K.

OR

- 9 (a) Derive Clausius clapeyron equation.
 - (b) 8 kg of steam at 8 bar and 0.8 dry is heated at constant pressure till the volume is doubled. Determine final quality of steam, heat added and change in internal energy.

UNIT – V

- 10 (a) 4 kg of O_2 at a pressure 100 kPa and 75°C are mixed with 7 kg of N_2 at the same pressure and temperature. Find the increase in entropy.
 - (b) A vessel of 0.5 m³ capacity contains 3 kg of CO₂ and 2 kg of N₂ at 320 K. Determine: (i) Pressure in the vessel. (ii) Mole fraction. (iii) R and M of the mixture.

OR

11 In an air standard Otto cycle the pressure at the beginning and end of compression are 1 bar, 15 bar the maximum pressure is 30 bar. Determine: (i) Compression ratio. (ii) Thermal efficiency. (iii) Mean effective pressure.

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