R13

Code: 13A03301

B.Tech II Year I Semester (R13) Supplementary Examinations June 2017

MATERIAL SCIENCE & ENGINEERING

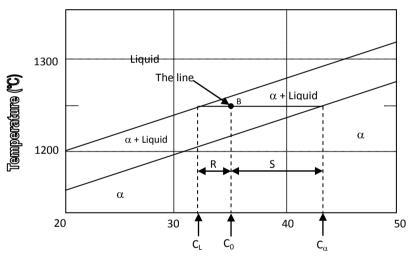
(Mechanical Engineering)

Time: 3 hours Max. Marks: 70

PART - A

(Compulsory Question)

- 1 Answer the following: $(10 \times 02 = 20 \text{ Marks})$
 - (a) What is unit cell? List any two metallic structures with examples.
 - (b) If the atomic radius of A1 is 0.143 nm then, calculate the volume of its unit cell.
 - (c) What are different reactions in binary phase diagram.
 - (d) How Grain boundaries influence the ductility of materials.
 - (e) What is the difference between brass & bronze?
 - (f) Briefly explain why grey cast iron is brittle.
 - (g) What is the difference between hardness & hardenability?
 - (h) Explain the normalizing treatment.
 - (i) Compute the fractions of each of α and liquid phases of given phase diagram. Given C_0 = 35 wt% Ni, C_a = 42.5 wt% Ni, C_L = 31.5 wt% Ni.



Composition (wt% Ni)

(j) What are the advantages of fiber reinforcement in composites?

PART - B

(Answer all five units, $5 \times 10 = 50 \text{ Marks}$)

UNIT – I

2 Define APF (Atomic Packing Factor) and derive unit cell length & calculate APF for FCC, BCC if the radius of atom is 'R'.

OR

3 Explain different crystalline defects with neat sketch.

[UNIT – II]

4 Explain any three heat treatment processes in detail.

OR

5 Explain the time-temperature-transformation (TTT) characteristics of eutectoid steel.

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(UNIT - III)

Write about different types of cast iron and explain their properties?

OR

7 How does carbon influences the properties of iron? What are the different types of steels?

UNIT - IV

8 Explain any two manufacturing methods of FRP (Fiber Reinforced Plastics).

OR

9 What are the benefits of composite materials over the metals and alloys? What are cermets?

UNIT - V

10 Draw "Fe-Fe₃C" diagram. Explain different reactions that occur in Iron Carbon diagram.

OR

What are different non-ferrous alloys? Explain the alloys of copper and aluminium.
